Binary Compound Work

REMEMBER - in each formula the cation goes first. For each compound, write…..

1. Names of elements
2. Symbol of each element
3. Oxidation number for each element
4. Final Formula
5. Aluminum + Sulfur =
6. Oxygen + calcium =
7. Copper (II) + Sulfur =
8. Oxygen + Iron (II) =
9. Lead (IV) + Phosphorus =
10. Lithium + Sulfur =
11. Magnesium + Fluorine =
12. Copper (II) + Chlorine =
13. Oxygen + Boron =
14. Bromine + Magnesium =
15. Nitrogen + Potassium =
16. Oxygen + Silicon =
17. Iodine + Calcium =
18. Phosphorous + Strontium =
19. Barium + Nitrogen =
20. Fluorine + Lead (IV) =
21. Chromium (III) + Sulfur =
22. Manganese (IV) + Nitrogen =
23. Oxygen + Aluminum =
24. Strontium + Phosphorous =
25. Silicon + Chlorine =
26. Iodine + Vanadium (III) =
27. Nitrogen + Potassium =
28. Nitrogen + Zinc (IV) =
29. Beryllium + Phosphorous =