Name

Section 2–4 Chemical Reactions and Enzymes (pages 49–53)

TEKS FOCUS: 1A Safe lab practices; 1B Conserve resources; 2A Plan investigations; 2B Measure precisely; 2C Make inferences; 2D Communicate valid conclusions; 3A Analyze scientific explanations; 4B Cellular processes

This section describes what happens to chemical bonds during chemical reactions. It also explains how energy changes affect chemical reactions and describes the importance of enzymes.

Chemical Reactions (page 49)

1. What is a chemical reaction? _____

2. Complete the table about chemicals in a chemical reaction.

CHEMICALS IN A CHEMICAL REACTION

Chemicals	Definition
Reactants	
Products	

3. Chemical reactions always involve changes in chemical ______.

Energy in Reactions (page 50)

- 4. What is released or absorbed whenever chemical bonds form or are broken?
- 5. What do chemical reactions that absorb energy need to occur? ______
- 6. Chemists call the energy needed to get a reaction started the _____
- 7. Complete the graph of an energy-releasing reaction by adding labels to show the energy of the reactants, the energy of the products, and the activation energy.



Na	me Class Date
En	zymes (pages 51–52)
8.	What is a catalyst?
9. 10.	Proteins that act as biological catalysts are called What do enzymes do?
11.	From what is part of an enzyme's name usually derived?
En 12.	zyme Action (pages 52–53) The reactants of enzyme-catalyzed reactions are known as
13.	Why are the active site and the substrates in an enzyme-catalyzed reaction often compared to a lock and key?
14.	The binding together of an enzyme and a substrate forms a(an)
15.	How do most cells regulate the activity of enzymes?